



# Energy and exergy analysis of a cycle-skipping strategy in an HCCI engine fueled with natural gas

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## Abstract

Homogeneous charge compression ignition (HCCI) engines have attracted considerable interest due to its incorporation of features from both gasoline and diesel engines. In this study, the effects of cycle-skipping strategies on engine performance, efficiency, and emissions in a natural gas-fired HCCI engine were investigated experimentally. Experiments were conducted under constant engine speed, at 25, 50, and 75% load levels, in the Normal, 3 Normal-1 Skip (3N1S), 2 Normal-1 Skip (2N1S), and 1 Normal-1 Skip (1N1S) cycle modes. Emissions, fuel consumption, energy–exergy flow, thermal losses, irreversibilities, entropy generation, thermal, and exergy efficiencies were calculated based on the experimental data. Under 50% engine load, the NO<sub>x</sub> emission in the 3N1S operating mode was measured at 1594 ppm, whereas it increased by 52.5% to 2431 ppm in the 2N1S mode. The correlation between cycle-skipping tactics and thermal efficiency was determined to be contingent upon engine load. In Normal and 3N1S modes, thermal efficiency generally improves with elevated engine loads, while the 2N1S and 1N1S modes provide superior performance at low to medium loads. At a constant 50% load, heat efficiency in the 2N1S mode increased by 7.84% to reach 28.34% compared to the Normal mode. Additionally, thermodynamic analyses revealed that the 1N1S mode has the lowest entropy generation and the least irreversibility, at 0.021 kW/K. These results demonstrate that cycle-skipping strategies can be an effective tool for optimizing engine performance based on load.

**Keywords** Cycle skipping · HCCI · Natural gas · Energy · Exergy

## Introduction

Energy is fundamental to contemporary societal advancement and sustainability. Since the Industrial Revolution, technological and industrial improvements have resulted in a persistent rise in worldwide energy demand [1–3]. The effective use of energy resources has become as crucial an issue as their acquisition. Internal combustion engines (ICEs), characterized by their high energy density and versatility

in fuel usage, continue to be essential in the transportation and mobility sectors. Currently, these engines are extensively utilized in several applications, such as automotive technologies, agricultural machinery, power generating, and maritime transportation [4]. Internal combustion engines transform chemical energy into mechanical energy by the regulated ignition of fuel within the combustion chamber, providing the versatility to function with many fuel types based on their core operational principles. As a result, the use of alternative fuels into internal combustion engines has become increasingly common [5–9]. The fuel type utilized in an internal combustion engine markedly affects emission levels [9, 10], engine efficiency [11, 12], and torque production [13, 14]. Natural gas, a significant alternative fuel, is extensively utilized in engines due to its abundance and reduced exhaust emissions relative to other fuels [15–17].

In Xing et al. [18], the performance and emission effects of low-carbon alternative fuel options were investigated in a marine vessel powered by an ICE. The findings indicated that the use of hydrogen and ammonia—followed by biodiesel and bioethanol—represents a viable option for marine

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transportation. Jayabal [19] examined the impact of a biodiesel derived from sapota seed biodiesel (SSB) and hydrogen gas blend on the operational and emission behavior of a dual-fuel CI engine. The hydrogen injection improved heat efficiency and significantly reduced emissions. The study reported an increase of 14.62% in brake thermal efficiency (BTE) and a reduction of up to 80% in emissions such as CO<sub>2</sub>, CO, and HC. Kumar et al. [20] investigated the performance, combustion, and emission characteristics of alternative fuels produced by blending waste plastic oil (WPO) and waste cooking oil-based biodiesel (WCOB) in an internal combustion diesel engine. According to the results, the B20P20D60 blend yielded the highest efficiency and the lowest emissions among the tested fuel combinations. M'hamed et al. [21] conducted an experimental study on diesel–alcohol blends by mixing ethanol and methanol with diesel fuel at varying ratios in an ICE. The study assessed how these fuel blends impact engine performance and examined the engine's power output, emissions, and energy economy. The results indicated that incorporating ethanol and methanol into diesel fuel has the potential to reduce emissions while enhancing overall efficiency.

In light of current energy policies and sustainability objectives, ICE technologies are undergoing a substantial transformation, both in terms of engineering advancements and compliance with environmental regulations. While the rise of electric and hybrid vehicles is reshaping the future of propulsion systems, innovative strategies aimed at developing more efficient and environmentally friendly combustion processes are actively being pursued [19, 22]. Diesel CI and gasoline SI engines necessitate continuous innovation—not only in engine design, but also in ensuring compatibility with a variety of fuel types. HCCI engines are an innovative technology that combines the advantages of both gasoline and diesel ICEs. In HCCI engines, a homogeneous fuel–air mixture is auto-ignited through compression within the cylinder, eliminating the need for a spark plug or direct injection ignition systems [23, 24]. Zapata-Mina et al. [25] conducted an experimental study on a single-cylinder HCCI engine using fuel blends composed of fusel oil and diethyl ether in varying ratios (D40F60, D60F40, and D80F20). The results, obtained under different engine speeds and lambda values, revealed that increasing the proportion of diethyl ether led to a decline in performance, with the lowest exergy efficiency of 5.34% recorded for the D80F20 blend. Tunçer et al. [26] conducted an experimental study on a HCCI engine powered by natural gas and evaluated its performance against a standard SI engine. Based on the findings, it was observed that the fuel consumption of the HCCI engine was 2.92% higher than that of the SI engine. Furthermore, the NO<sub>x</sub> emission was measured at 19.84 g/kWh for the HCCI engine, whereas it was 35.72 g/kWh for the SI engine.

Verma et al. [27] investigated the effects of various fuels on engine performance and emissions in an HCCI engine. The results indicated that the use of natural gas in HCCI operation led to at least a 10% improvement in engine efficiency compared to diesel fuel. Additionally, it was found that HCCI engines produced lower exhaust emissions relative to conventional internal combustion engine types. Çelibi et al. [28] conducted an experimental study using different ratios of naphtha blends (N25, N50, and N75) and n-heptane in a single-cylinder HCCI engine with a maximum power output of 15 kW. At a lambda ( $\lambda$ ) value of 2, an intake air temperature of 60 °C, and an engine speed of 1400 rpm, the CO emission values for n-heptane and N75 fuels were measured as 0.048% and 0.08%, respectively. In HCCI engines, parameters such as fuel injection strategy [29–31] and compression ratio [32, 33] significantly influence engine performance and have been the subject of numerous studies.

Cycle-skipping strategy (SCS) is a method used to reduce the energy losses of the engine under low load conditions [34]. This approach dynamically regulates the effective stroke volume by deactivating specific engine valves, thereby minimizing pumping losses [35, 36]. As a result, SCS allows the engine to operate with lower energy consumption under partial load conditions, enhancing overall engine efficiency [36]. There are studies examining the effects of SCS on engine performance and emissions in various engine types [37–39]. Tunçer [40] experimentally evaluated the use of pure diesel and pure natural gas (NG100) in a single-cylinder engine under normal operation and two different SCS modes (2N1S and 3N1S) at a fixed speed of engine of 1500 rpm. Exergy and energy analyses were performed using the data. At 75% engine load, the heat efficiency was calculated as 34% in the natural gas-fueled normal cycle and 30% in the 2N1S cycle. Under 25% engine load, the exergy destruction in the 2N1S and 3N1S modes was found to be 13 kW and 16.54 kW, respectively.

The use of the skip cycle system in a four-stroke, single-cylinder, water-cooled spark ignition engine running at a steady 2000 rpm was empirically examined by Kutlar et al. [34]. The results indicated that rising the engine speed led to a significant rise in fuel consumption under cycle-skipping conditions. At 2000 rpm, the brake-specific fuel consumption (BSFC) was recorded as 680 g/kWh in the Normal mode of operation, whereas it increased to 954 g/kWh in the normal–skip–skip (NSS) mode. Tunçer et al. [41] conducted tests under three different operating conditions—Normal, 3N1S, and 2N1S—in a single-cylinder, natural gas-fueled SI engine using the skip cycle method. According to the CO emission data, the lowest CO emission was recorded under the normal operation mode, while the highest CO emission, 116 g/kWh, was recorded in the 2N1S mode. Additionally,

the engine exhibited its highest heat efficiency, 28%, under the 3N1S operating condition.

Various studies in the literature have concentrated on the performance, emissions, and thermodynamic assessments of HCCI engines. Nonetheless, there is a significant deficiency in research concerning exergy and energy studies for HCCI engines utilizing natural gas fuel under cycle-skipping conditions.

This study reveals how cycle-skipping modes in natural gas-fueled HCCI engines create a difference not only in performance but also in thermodynamic terms, and in this respect, expands the depth of technical analysis in the literature.

## System components and application methodology

### Data preparation

This paper involves the modification of a typical diesel engine to utilize natural gas as its primary fuel. The original diesel injector was extracted, and a spark plug was fitted into the cylinder head to enable ignition. A throttle valve and natural gas injectors were custom-manufactured and built into the system to control gas flow. Inductive sensors were fitted on the engine to measure crankshaft velocity and to ascertain the position of the top dead center (TDC). The data obtained from these sensors were processed using computer programs that are specialized.

Engine mapping revealed that the highest engine load occurred at 1500 rpm. Consequently, tests were conducted at three distinct engine loads: 75, 50, and 25%. Cycle-skipping strategies were applied by modifying ignition timing and gas injection patterns. Four operating modes were tested: Normal (N), 3 Normal–1 Skip (3N1S), 2 Normal–1 Skip (2N1S), and 1 Normal–1 Skip (1N1S). A schematic diagram of the experimental setup is presented in Fig. 1, while the technical specifications are summarized in Tables 1 and 2.

For HCCI operation, the intake air was preheated to raise the in-cylinder temperature, enabling the natural gas to reach its auto-ignition point and achieve stable combustion. Due to its higher auto-ignition temperature compared to gasoline and diesel, natural gas required intake air temperatures above 100 °C to ignite spontaneously. To ensure consistent thermal conditions during the tests, two PT100 temperature sensors were installed—one at the intake port and one before the intake valve.

Additionally, a knock sensor was mounted to monitor and prevent engine knocking. When switching to HCCI mode, the spark plug cable was disconnected to disable spark ignition. The engine was allowed to stabilize under HCCI conditions before initiating the test measurements.

During the experimental studies, signals corresponding to TDC position, in-cylinder pressure, gas injection, and ignition were recorded using an oscilloscope. Emission measurements were conducted using a Bosch BEA 060 gas analyzer in accordance with the ISO 3046-1 standard. The overall experimental flowchart is illustrated in Fig. 2.

### Error and uncertainty analysis

Ensuring the reliability of the study requires the verification of data obtained from engine tests. Measurement errors may arise due to factors such as instrument selection, ambient conditions, reading inaccuracies, and calibration-related uncertainties during the experiments. In this study, an uncertainty analysis was conducted to identify and minimize potential measurement errors. The accuracy levels and uncertainty values associated with the measurements are presented in Table 3.

### Energy analysis

An exergy and energy analysis of an ICE was conducted to investigate its performance and emission characteristics under different operating modes is presented in Fig. 3. In this analysis, the engine is assumed to operate under steady-state conditions, and the following assumptions are applied:

- Combustion air and exhaust gases are considered ideal gases.
- Average values are used for specific heat capacities.
- Changes in kinetic and potential exergies are neglected.
- The ambient temperature and pressure are taken as 293 K and 1 atm, respectively.

The mass flow rate into the system ( $\dot{m}_{in}$ ) is equal to the mass flow rate out of the system ( $\dot{m}_{out}$ ) for a control volume with steady flow, as stated in Eq. (1), in accordance with the principle of mass conservation [45].

$$\sum \dot{m}_{in} = \sum \dot{m}_{out} \quad (1)$$

The difference between the energy transported by the mass entering and leaving the system equals the difference between the heat transferred to the system ( $\dot{Q}$ ) and the work produced by the system ( $\dot{W}$ ) [46]. The energy balance is expressed by Eq. (2).

$$\sum \dot{m}_{out} h_{out} - \sum \dot{m}_{in} h_{in} = \dot{Q} - \dot{W} \quad (2)$$

General representation of the energy conservation law for control volumes with multiple inlets and outlets is given in Eq. (3). The rate of chemical energy supplied to the engine via fuel can be calculated using Eq. (4), where  $H_u$  is the

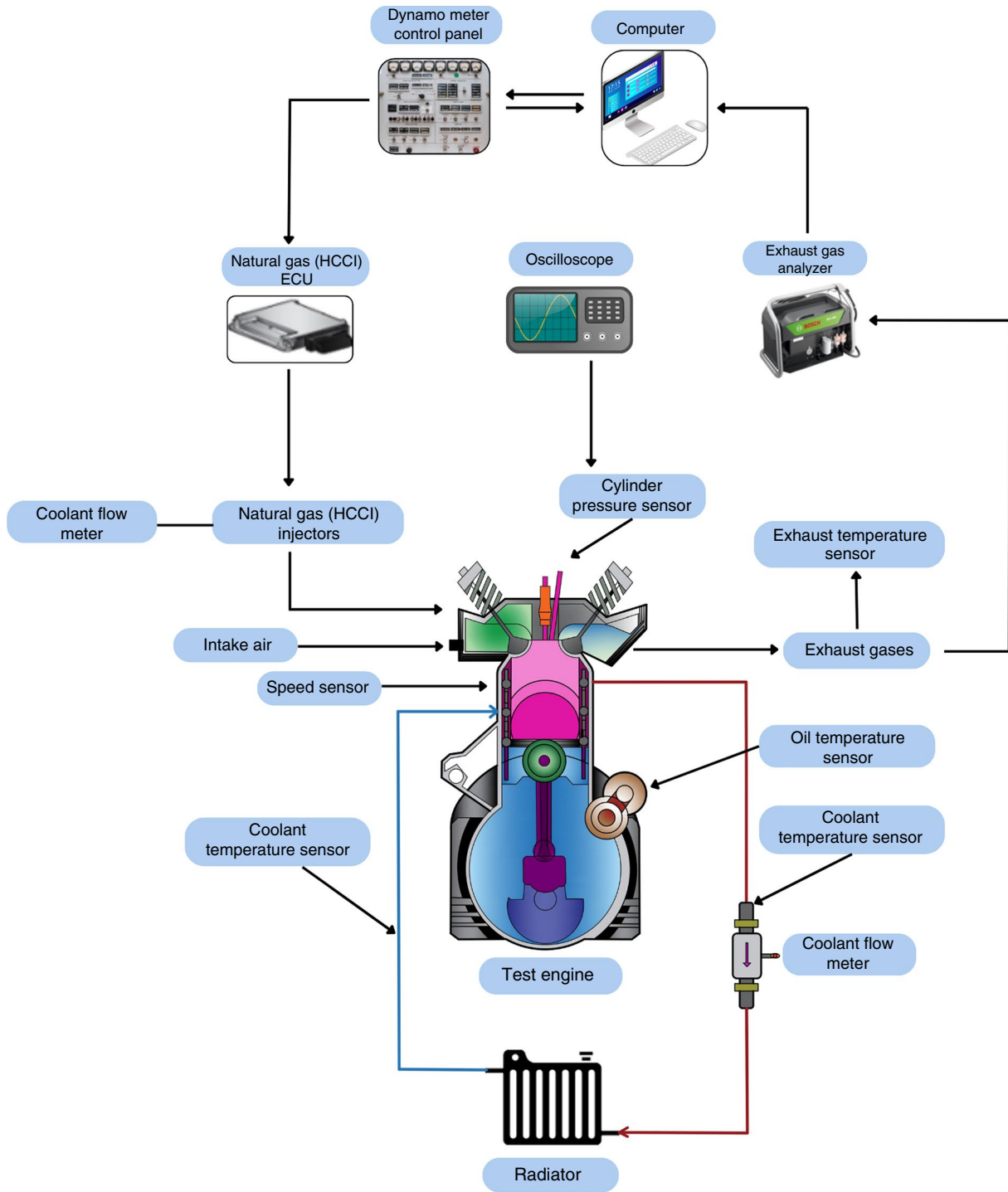


Fig. 1 Experimental setup block diagram

lower heating value of the fuel and  $\dot{m}_{\text{fuel}}$  is the mass flow rate of the fuel [47].

$$\dot{Q}_{\text{in}} - \dot{Q}_{\text{out}} + \sum \dot{m}_{\text{in}} h_{\text{in}} = \dot{W}_{\text{out}} - \dot{W}_{\text{in}} + \sum \dot{m}_{\text{out}} h_{\text{out}} \quad (3)$$

$$\dot{E}_{\text{fuel}} = \dot{m}_{\text{fuel}} H_u \quad (4)$$

In an engine control volume, the total chemical energy of the fuel ( $\dot{E}_{\text{fuel}}$ ) is either converted into mechanical work ( $\dot{W}$ ) or dissipated as heat losses to the environment ( $\dot{Q}_{\text{loss}}$ ). The relationship is expressed by Eq. (5) [48].

**Table 1** Technical characteristics of the engine used in experimental studies

Parameters	Specification
Engine supplier	Erin Engine Corporation Inc
Brand	Erin Engine
Model	1.16L Diesel—1.16L Natural gas engine
Cylinder number	1
Engine cycle	4
Maximum engine power	10 kW @ 1500 rpm
Maximum engine torque	60 Nm @ 1500 rpm
Powertrain	Camshaft located in the engine with push-rod mechanism
Valve system	4 valves per cylinder
Fuel injection type	Port injection (Natural gas)
Ignition	Compression-ignition (Diesel) Electronically Spark-ignition (Natural gas)
Cooling system	Water cooled
Idle speed	1500 rpm
Swept volume	1160 cm <sup>3</sup>
Cylinder bore	108 mm
Stroke	127 mm
Compression ratio	14.6:1
Number of injectors nozzle	6
Intake valve timing advance	25° BTDC
Intake valve closure retardation	65° ABDC
Injection timing of fuel	19° BTDC
Opening advance of exhaust valve	55° BBDC
Closing delay of exhaust valve	35° ATDC

**Table 2** Technical specifications of fuels used in the experimental study [42–44]

Property	Natural gas
Fuel density at 15 °C/kg m <sup>-3</sup>	0.72–0.75
Fuel viscosity/mm <sup>2</sup> s <sup>-1</sup>	16.33
Cetane number	–
Octane number	120
Molecular weight/kg mol <sup>-1</sup>	16.04
Lower calorific value/MJ kg <sup>-1</sup>	46.0–48.0
Latent heat of evaporation/kJ kg <sup>-1</sup>	511
Percentage of methane by volume	92.5
Percentage of ethane by volume	3
Percentage of propane by volume	0.71
Percentage of butane by volume	0.02
Percentage of pentane by volume	0.1
Percentage of carbon dioxide by volume	0.6
Percentage of nitrogen by volume	3

$$\dot{Q}_{\text{loss}} = \dot{E}_{\text{fuel}} - \dot{W} \quad (5)$$

The heat efficiency ( $\eta_{\text{th}}$ ) is a key indicator of the energy conversion performance of the system. It represents the ratio

of useful work output to the total fuel energy input and is given by Eq. (6) [47].

$$\eta_{\text{th}} = \frac{\dot{W}}{\dot{E}_{\text{fuel}}} \quad (6)$$

### Exergy analysis of the system

Exergy analysis is a thermodynamic approach employed to detect irreversibilities in a system and evaluate the potential for recoverable energy, facilitating more efficient energy transfer procedures. This analysis focuses on the quality of energy and the amount of usable energy. The exergy balance for the control volume of the engine is expressed by Eq. (7), where the incoming ( $\dot{E}x_{\text{in}}$ ) and outgoing ( $\dot{E}x_{\text{out}}$ ) exergy flow rates are equal under ideal, reversible conditions [49].

$$\dot{E}x_{\text{in}} = \dot{E}x_{\text{out}} \quad (7)$$

In this study, the exergy of incoming air ( $\dot{E}x_{\text{air}}$ ) is assumed to be zero, as the air is considered to enter the engine under ambient atmospheric conditions. The exergy of the fuel ( $\dot{E}x_{\text{fuel}}$ ) is distributed into work exergy ( $\dot{E}x_{\text{work}}$ ), exhaust gas

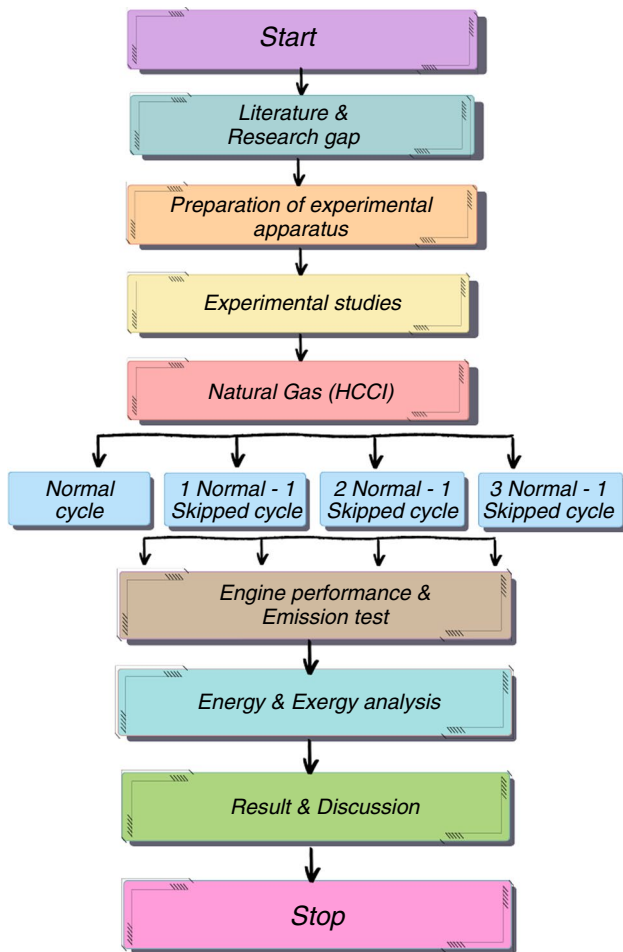


Fig. 2 General flowchart of the this study

Table 3 Inaccuracy in measured data and computational results

Parameter	Measuring range	Accuracy	Uncertainty/%
The power of engine	–	–	±0.57
The speed of engine	0–2000 rpm	± 1 rpm	±0.063
BSFC	–	–	±0.762
Flow rate of Fuel	–	–	±0.41
Flow rate of Air	–	–	±0.52
Brake torque	0–50 Nm	±0.1 Nm	±0.50
Temperature	-	±0.1 °C	± 1.0
NO	0–5000 ppm	± 1 ppm	±0.601
O <sub>2</sub>	0–25% vol	±0.01%	±0.561
CO	0–5% volume	±0.001%	±0.823
HC	0–3000 ppm	± 1 ppm	±0.925

exergy ( $\dot{E}x_{eg}$ ), heat losses ( $\dot{E}x_{heat}$ ), and irreversibilities ( $\dot{E}x_{dest}$ ) as expressed in Eq. (8) [48].

$$\dot{E}x_{air} + \dot{E}x_{fuel} = \dot{E}x_{work} + \dot{E}x_{eg} + \dot{E}x_{heat} + \dot{E}x_{dest} \quad (8)$$

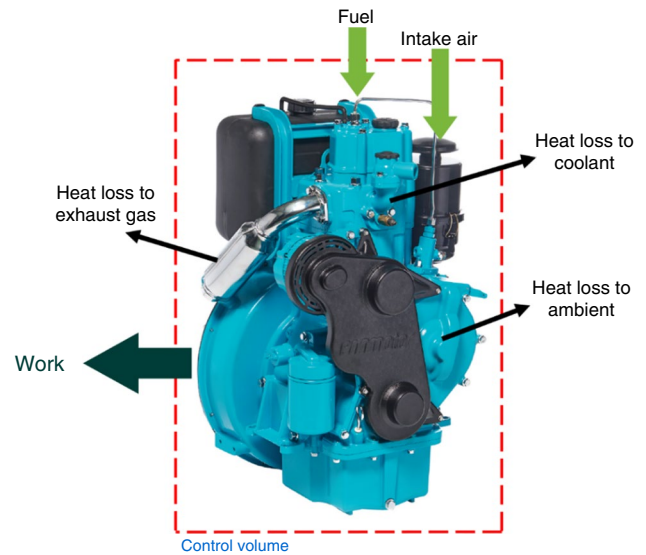


Fig. 3 Engine used in experimental studies

The elemental composition of the natural gas used in this study is as follows: hydrogen (H) 12.982%, carbon (C) 86.984%, and sulfur (S) 0.034%. Based on this composition, the exergy factor ( $\varphi$ ) of the fuel is calculated using Eq. (9) [50].

$$\varphi = 1.033 + 0.0169\left(\frac{n}{m}\right) - \left(\frac{0.0698}{m}\right) \quad (9)$$

The specific exergy of the fuel ( $\epsilon_{fuel}$ ) is obtained by multiplying the lower heating value of the fuel ( $\dot{E}x_{fuel}$ ) by its exergy factor [51]:

$$\epsilon_{fuel} = H_u \varphi \quad (10)$$

$$\dot{E}x_{fuel} = \dot{m}_{fuel} \epsilon_{fuel} \quad (11)$$

The total exergy of the exhaust gases at the engine outlet consists of two components: physical exergy ( $\epsilon_{phy}$ ) and chemical exergy ( $\epsilon_{chem}$ ).  $\epsilon_{phy}$  is derived from the derivations in enthalpy and entropy from reference environment, while chemical exergy reflects the chemical potential of gas mixture based on its composition. The calculations are based on mole fractions ( $y^e$ ) of the reference value at ambient temperature ( $T_0$ ) as shown in Table 4. The total exergy of the exhaust gas is the sum of its physical and chemical exergy components and represents the overall exhaust exergy potential [52, 53].

$$\epsilon_{phy} = [(h - T_0s) - (h_0 - T_0s_0)] \quad (12)$$

$$\epsilon_{chem} = RT_0 \ln \frac{1}{y^e} \quad (13)$$

**Table 4** The elements of the environment [54]

Reference component	Mol fractions ( $y^e$ ) /%
N <sub>2</sub>	75.6700
O <sub>2</sub>	20.3500
CO <sub>2</sub>	0.03450
H <sub>2</sub> O	3.03000
CO	0.00070
SO <sub>2</sub>	0.00020
H <sub>2</sub>	0.00005
Others	0.91455

$$\varepsilon = \varepsilon_{\text{phy}} + \varepsilon_{\text{chem}} \quad (14)$$

The exergy loss due to heat dissipation in an engine ( $\dot{E}x_{\text{heat}}$ ) is a parameter that depends on the ambient temperature, the system temperature ( $T_s$ ), and the rate of heat transfer to the environment via the cooling water ( $\dot{Q}_{\text{cw}}$ ). This quantity is calculated to evaluate the heat losses occurring in the system. The exergy associated with heat transfer is determined using Eq. (15) [55]. The amount of heat dissipated by the cooling water ( $\dot{Q}_{\text{cw}}$ ) is calculated using Eq. (16), based on the difference between the total fuel energy input and the sum of mechanical work output and enthalpy change of the exhaust gases.

$$\dot{E}x_{\text{heat}} = \sum \left( 1 - \frac{T_0}{T_s} \right) \dot{Q}_{\text{cw}} \quad (15)$$

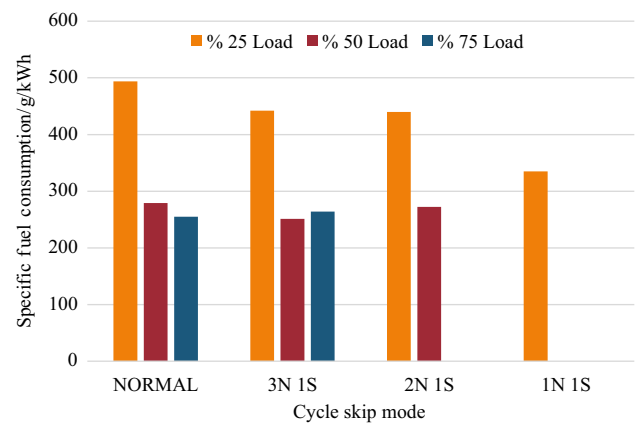
$$\dot{Q}_{\text{cw}} = \dot{m}_{\text{fuel}} H_u - (\dot{E}x_w + \dot{m}_{\text{out}} \Delta h_{\text{out}}) \quad (16)$$

Exergy destruction ( $\dot{E}x_{\text{dest}}$ ) plays a critical role in evaluating system inefficiencies, as it quantifies the non-recoverable energy losses. It is calculated as the difference between the total exergy input from the fuel and the sum of useful work output, exhaust exergy, and heat loss exergy, as given in Eq. (17). Entropy generation ( $\dot{s}_{\text{gen}}$ ) is a measure of the irreversibility and disorder introduced into the system. It is calculated by dividing the  $\dot{E}x_{\text{dest}}$  by the ambient temperature, as shown in Eq. (18) [56].

$$\dot{E}x_{\text{dest}} = \dot{E}x_{\text{fuel}} - (\dot{E}x_{\text{work}} + \dot{E}x_{\text{eg}} + \dot{E}x_{\text{heat}}) \quad (17)$$

$$\dot{s}_{\text{gen}} = \frac{\dot{E}x_{\text{dest}}}{T_0} \quad (18)$$

Exergy efficiency ( $\psi$ ), also referred to as the second law efficiency, is a key performance indicator for assessing irreversibilities and entropy production within the system. It is defined as the ratio of the useful exergy output (work) to the total exergy input and is expressed in Eq. (19) [57].


**Fig. 4** Fuel consumption in HCCI engine at different engine loads and operating modes

$$\psi = \frac{\dot{E}x_w}{\dot{E}x_{\text{in}}} \quad (19)$$

A high exergy efficiency indicates that a greater portion of the input fuel exergy is effectively converted into useful work. Conversely, a low exergy efficiency suggests that a significant amount of energy is lost to the environment or consumed in irreversible processes.

## Experimental results

Performance, emission, and thermodynamic analyses (energy and exergy) were conducted on a natural gas-fueled HCCI engine operating under four distinct cycle-skipping strategies (Normal, 3N1S, 2N1S, and 1N1S) and at three engine loads (25, 50, and 75%). Critical parameters such as fuel consumption, exhaust emissions, heat losses, energy and exergy flows, irreversibilities, entropy generation, heat efficiency, and exergy efficiency were evaluated based on experimental data.

Fuel consumption, which represents the mass of fuel consumed per unit output energy, is a crucial indicator for evaluating both system efficiency and environmental impact. As shown in Fig. 4, fuel consumption generally decreased with increasing engine load. For instance, in 2N1S mode, fuel consumption decreased from 439.97 g kWh<sup>-1</sup> at 25% load to 272.46 g kWh<sup>-1</sup> at 50% load, reflecting a 38.07% reduction. At 25% load, fuel consumption in Normal and 3N1S modes was 493.93 g kWh<sup>-1</sup> and 442.24 g kWh<sup>-1</sup>, respectively.

A decrease in fuel consumption is observed as the engine load increases. In the 2N1S operating mode, while the fuel consumption was 439.97 g kWh<sup>-1</sup> at 25% engine load, this value decreased by 38.07% to 272.46 g kWh<sup>-1</sup> at 50% engine load. The reason for the deterioration of the general trend in

3N1S and 2N1S operating modes with engine load change is that more fuel consumption is needed to obtain the same torque. Yüksel et al. [35] studied the effect of different cycle-skipping strategies on engine performance and emissions using an experimental approach. The results show that a 4.3% reduction in fuel consumption is achieved in cycle-skipping strategies compared to normal mode.

Table 5 shows the measured values of carbon monoxide (CO), hydrocarbons (HC), and nitrogen oxides (NO<sub>x</sub>) emission values obtained for different engine loads and operating modes. In the study, the emissions from normal operation mode and different operating modes such as 3N1S, 2N1S, and 1N1S were compared. Emission measurements are essential to determine the effect of various operating modes and engine loads on exhaust gas components. Engine load is a primary factor influencing the combustion process, thereby affecting emission characteristics. At lower loads, combustion temperatures are reduced, typically resulting in lower NO<sub>x</sub> but potentially higher HC and CO emissions. Conversely, at higher loads, elevated temperatures and pressures intensify NO<sub>x</sub> formation.

The highest CO emission was calculated as 0.534% in the 3N1S operating mode. HC emission was recorded at 128 ppm in the 2N1S mode, whereas it decreased by 2.88% to 91 ppm in the 1N1S mode at a steady engine load of 25%. Increasing engine load generally led to an increase in NO<sub>x</sub> emissions. In normal mode, raising the load to 50% resulted in a NO<sub>x</sub> level of 1248 ppm. Baykara et al. [58] investigated the impact of the cycle-skipping strategy on fuel consumption and exhaust emissions under varying engine speeds and operating pressures. The study revealed that the implementation of the cycle-skipping strategy led to a marked increase in NO<sub>x</sub> emissions, which is considered an unfavorable outcome in terms of environmental impact.

**Table 5** Emission values in the engine at different engine loads and operating modes

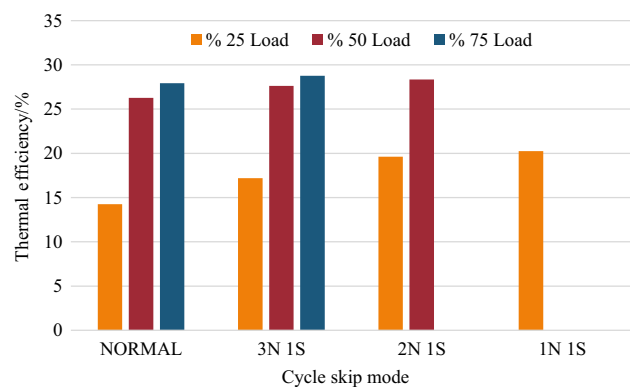
Cycle-skip mode	Engine load/%	CO/%	HC/ppm	NO <sub>x</sub> /ppm
Normal	25	0.018	105	13
	50	0.028	89	72
	75	0.036	115	1320
03N 1S	25	0.026	97	56
	50	0.011	94	1594
	75	0.534	397	1678
2N 1S	25	0.024	128	86
	50	0.051	236	2431
	75	0	0	0
1N 1S	25	0.033	91	2461
	50	0	0	0
	75	0	0	0

Energy flow and heat losses, which are important indicators of engine performance, were calculated for each engine load and operating mode and are summarized in Table 6. In both normal and alternating modes (3N1S, 2N1S, 1N1S), energy input to the engine increased with increasing load. While more energy is generated at higher loads, a portion is inevitably dissipated as heat to the surroundings. Doubling the engine load in normal and 2N1S modes led to increases in thermal losses by 0.944% and 12.006%, respectively. The effect of load on energy flow was more significant in skip modes than in normal mode. In 3N1S operation, the difference in thermal losses between maximum and minimum loads was calculated as 11.052 kW.

Thermal efficiency, based on the first law of thermodynamics, quantifies the portion of input fuel energy converted into mechanical work. High thermal efficiency indicates effective utilization of fuel energy with minimal losses. Figure 5 illustrates the variation of thermal efficiency

**Table 6** Energy analysis of the engine at different engine loads and operating modes

Cycle-skip mode	Engine load/%	Energy flow/kW	Heat loss/kW
Normal	25	18.948	14.83
	50	21.309	14.97
	75	27.576	18.15
3N 1S	25	15.713	13.01
	50	20.277	14.91
	75	26.765	19.98
2N 1S	25	13.758	12.91
	50	19.761	14.46
	75	0.000	0.00
1N 1S	25	13.331	9.17
	50	0	0
	75	0	0



**Fig. 5** Thermal efficiency of the engine under varying loads and operational modes

with engine load and operating mode. In normal and 3N1S modes, thermal efficiency increased with load, while 2N1S and 1N1S modes exhibited better performance at low and medium loads. The 2N1S mode is an intermediate strategy that involves two combustion cycles in three. This mode provides a more consistent operation compared to the 1N1S mode. It can meet the required torque, particularly under 50% engine load, while also protecting personnel by reducing pumping losses. This improves combustion stability and achieves efficient power generation.

Both strategies, thanks to cycle-skipping options, allow for complete exhaust gas evacuation during non-combustion cycles and the preparation of fresher mixtures in the next cycle. This contributes to an increase in combustion temperature and exergy temperature. At 25% engine load, thermal efficiency was 14.24% and 20.25% in the normal and 1N1S modes, respectively. The peak thermal efficiency, 28.77%, was recorded at 75% load in the 3N1S mode.

Exergy analysis serves as a diagnostic tool to assess the deviation of the system from ideal thermodynamic behavior, primarily by quantifying irreversibilities. This analysis involves calculating exergy input, exhaust exergy, and thermal exergy losses. The second law (exergy) efficiency, which reflects the quality of energy conversion, was also determined in this study and is presented in Table 7. At higher engine loads, the system releases more heat and when this heat is not distributed evenly in the engine or cannot be recovered effectively, irreversibilities due to heat transfer increase. In the 3N1S mode, the fuel exergy at 75% engine load was found to be 70.37% higher than at 25% engine load.

Exergy flow rates were generally lower under cycle-skipping conditions compared to normal mode. This situation is due to the change in the amount of fuel injected in order to obtain the same torque in the cycle-skip modes. The exergy associated with exhaust gases represents the portion

of energy that is unavailable for conversion into useful work. An increase in exhaust exergy was observed under skip modes compared to normal mode.

The cycle-skipping strategy positively influenced combustion by reducing both exhaust and thermal exergy. At 25% engine load, exhaust exergy in the normal mode was 4.85 kW, while it decreased by 9.734% to 4.422 kW in the 1N1S mode. The highest thermal exergy, 3.026 kW, was recorded in the normal mode at 75% load. Conversely, at 25% load, thermal exergy in the 1N1S mode was 49.34% lower than in the normal mode, mainly due to lower exhaust gas temperatures, as illustrated in Fig. 6.

Exergy destruction in the engine results from irreversibilities occurring during energy conversion processes. Minimizing exergy destruction is critical for achieving maximum performance and reducing emissions from a thermodynamic terms. As engine load increases, a greater amount of fuel is introduced into the control volume, leading to higher energy demand. Consequently, fuel consumption rises,

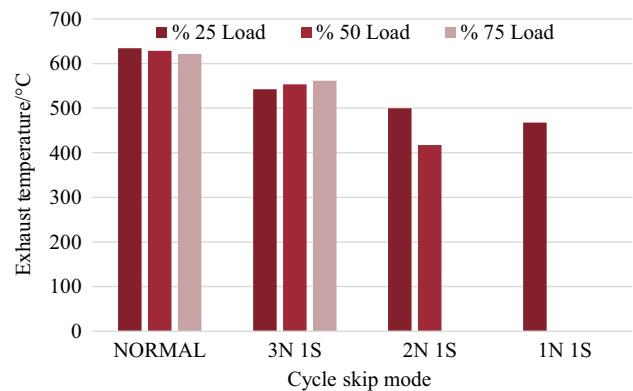


Fig. 6 Exhaust temperatures of the engine under varying loads and operational modes

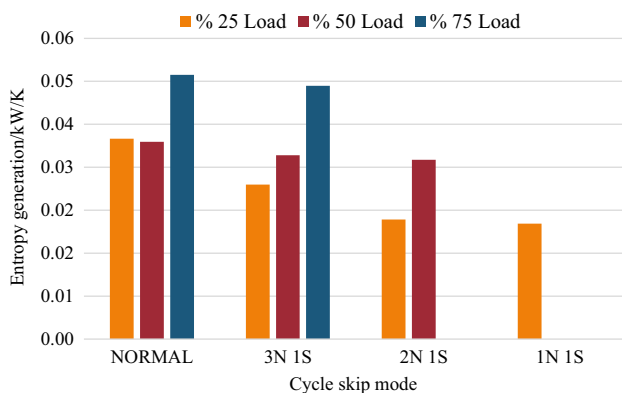
Table 7 Exergy analysis of the engine at different engine loads and operating modes

Cycle-skip mode	Engine load/%	Exergy flow kW	Exhaust exergy/kW	Heat exergy/kW	Exergy destruction/kW
Normal	25	20.653	4.859	1.893	11.201
	50	23.227	4.703	1.903	11.020
	75	30.058	4.565	3.026	14.767
3N 1S	25	17.127	4.666	1.134	8.627
	50	22.102	4.546	1.688	10.268
	75	29.174	4.424	2.893	14.157
2N 1S	25	14.996	4.530	1.086	6.680
	50	21.539	4.396	1.521	10.022
	75	0	0	0	0
1N 1S	25	14.531	4.422	0.959	6.449
	50	0	0	0	0
	75	0	0	0	0

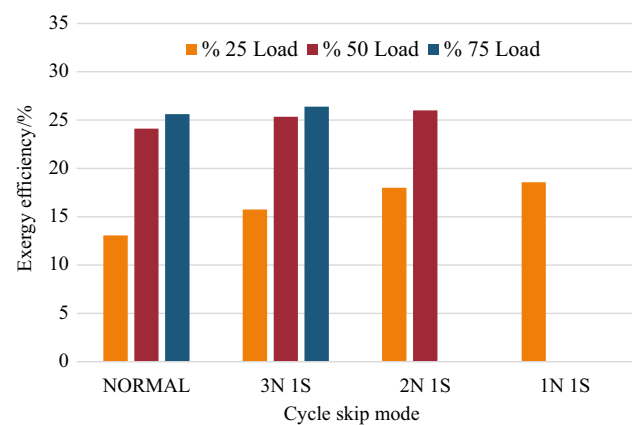
and thermodynamic irreversibilities become more prominent, thereby increasing exergy losses. In the 3N1S mode, exergy destruction increased from 10.27 kW at 50% load to 14.16 kW at 75% load. At 25% load, the 1N1S mode exhibited 42.41% less exergy destruction compared to the normal operating mode. The highest exergy destruction recorded in the study was 14.77 kW, observed in the normal mode at 75% engine load.

Entropy production and irregularities occurring during the engine operation process cause irreversibility. According to the second law of thermodynamics, every real system produces some entropy, causing energy quality to decrease. Higher entropy production causes more irreversibility and lower engine performance. Therefore, minimizing entropy production is an important issue for maximum engine efficiency. The entropy produced depending on the change of operating modes and engine loads is given in Fig. 7. The 1N1S mode is a strategy that only performs combustion once in every two engine cycles; in this case, combustion events occur once at a crankshaft angle of  $1440^\circ$ . When this strategy is implemented using the HCCI principle, the frequency of combustion decreases and average in-cylinder temperatures decrease. By reducing the frequency of combustion events, temperature changes between successive cycles are limited, reducing thermal gradients and instability in the system. In this study, the 1N1S mode was observed to have the lowest entropy generation at 0.021 kW/K. The 3N1S mode has 4.065% less entropy production at 75% engine load compared to the normal mode without cycle-skip mode.

In the energy conversion stages, not all of the energy can be converted into useful work and some of it loses its usability. According to the second law of thermodynamics, all real processes produce entropy, and increased entropy generation indicates greater thermodynamic inefficiencies and reduced engine performance. Therefore, exergy efficiency represents the quality of energy utilization within the system. In this study conducted on a natural gas-fueled



**Fig. 7** Entropy production in the engine under varying loads and operational modes



**Fig. 8** Exergy efficiency of the engine under varying loads and operational modes

engine, the second law efficiency values for both the normal and alternative operating modes are presented in Fig. 7. The highest exergy efficiency was obtained in the 3N1S operating mode at 75% engine load, with a calculated value of 26.39% is presented in Fig. 8. An improvement in exergy efficiency was observed when transitioning from the normal operating mode to cycle-skipping modes. At an engine load of 25%, the exergy efficiency in the normal operating mode was calculated as 13.073%, whereas this value increased to 18.581% in the 1N1S mode.

## Conclusion and discussion

This study conducted performance, emission, exergy, and energy evaluations utilizing experimental data from tests carried out at varying engine loads (25, 50, and 75%) and under distinct operating modes (Normal, 3N1S, 2N1S, and 1N1S) in a natural gas-fueled HCCI engine. The environmental impact of the engine was evaluated by measuring critical emission characteristics, including unused hydrocarbons (HC), nitrogen oxides ( $\text{NO}_x$ ), and carbon monoxide (CO). It was observed that irreversibility increased significantly at high engine loads when the generated heat could not be effectively recovered. In the 3N1S mode, thermal exergy at 75% engine load was 154.97% higher compared to its value at 25% engine load. The maximum fuel exergy was determined as 30.058 kW at 75% engine load under the normal operating mode. Second law (exergy) efficiency, which is one of the most critical indicators of engine performance, increased with rising engine load. While the exergy efficiency in the normal mode at 25% engine load was calculated as 13.073%, this value increased to 18.581% in the 1N1S mode, where the cycle-skipping strategy was applied.

According to the findings, at low engine loads, the 1N1S mode offers a more balanced and optimized operation in

terms of exergy, while at high loads, the 3N1S mode stands out as a more suitable option for increasing system efficiency. This situation highlights the need to develop hybrid strategies based on engine load. Furthermore, integrated cooling systems or waste heat utilization technologies could be considered for future studies to reduce irreversibilities in situations where heat recovery potential is high. Future research is recommended to utilize and optimize different alternative fuels in conjunction with cycle-skipping strategies.

## Limitations

The effectiveness of skip cycle strategies may vary at different engine speeds. Therefore, during transient conditions (such as sudden acceleration or load increases), delayed ignition, incomplete combustion in the cylinder, or misfires may occur. Evaluating these effects should be addressed in future studies.

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